## P P SAVANI UNIVERSITY

Third Semester of B.Sc. Examination December-2021

## SSCH2011-Inorganic Chemistry-II

07.12.2021, Tuesday

Time: 09:00 am to 11:30 am

Maximum Marks: 60

7		-			40			
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- 1. The question paper comprises of two sections.
- 2. Section I and II must be attempted in separate answer sheets.
- 3. Make suitable assumptions and draw neat figures wherever required.
- 4. Use of scientific calculator is allowed.

## Section-I (Total Marks - 30)

Q.1 Very Short Questions

(10)

1.1 Objectives

05

- 1.1a Which metal plays an important role in muscle contraction?
  - A Aluminum
  - B Sodium
  - C Magnesium
  - D Calcium
- **1.1b** Siderophores are responsible for
  - A Iron transport
  - B Magnesium transport
  - C Iron synthesis
  - D Magnesium synthesis
- 1.1c The ferritin has not been combined with iron is commonly termed as.
  - A aboferritin
  - B acoferritin
  - C apoferritin
  - D adoferritin
- 1.1d Which is super paramagnetic protein?
  - A Caretin
  - B Hemosiderin
  - C Ferritin
  - D globulin
- 1.1e Which metal important for the production of Globular protein?
  - A Ferrous
  - **B** Magnesium
  - C Vanadium
  - **D** Calcium
- 1.1f Full form of FDA
  - A Food and Drug Administration
  - B Foam and Drug Administration
  - C Food and Dye Administration
  - D Food and Drug Advertisement

1.1	Which Chelator used in gold toxicity?  A DMSA B Deferoxamine C DMPS D Penicillamine	E
1.1	A Amino peptidase B Phosphatase C Carboxypeptidase D Oxidase	
1.1 1.1j	A H <sub>2</sub> O B NH <sub>3</sub> C NO+ D CO Who classified the complex on the basis of their stability in solution? A Bilzt	
1.2	Caostions Hill (U/3) Hill Hill fin the Di	
1.2b	Give the name of Toxic metals Give the name of K <sub>2</sub> [PdCl <sub>4</sub> ] Give the use of Cadmium metal	05
Q.2 A B C	Short Notes (Attempt any two) Coordination Polymer Polydentate Ligand Infra-red Techniques for distinguish for trans-isomers. Importance of Polynuclear complex	06
Q.3 A B	Explain in detail (Attempt any two) Explain Ferritin What is Stereo Isomerism? Explain Geometrical isomerism in 4 Co-ordination Compound. Classification of Complex compound.	14
	Section-II (Total Marks - 30)	
Q.1 l.1 l.1a	Very Short Questions Objectives Complex K4[Fe (CN)6] nomenclature is A Potassium hexacyanoferrate (II) B Potassium hexacyanoferrate (I) C Potassium cyanohexaferrate (II) D Potassium hexacyanoferrate (III)	(10) 05-

1.1b	Complex K <sub>2</sub> [OsCl <sub>5</sub> N] nomenclature is  A Potassium Chloropentanitridoosmate (2 <sup>-</sup> )  B Potassium Pentachloronitridomate (2 <sup>-</sup> )  C Potassium Pentachloronitridoosmate (2 <sup>-</sup> )  D Potassium Chloropentanitridoosmate (2 <sup>-</sup> )
1.1c	What is isomer?  A Same chemical composition and Same structural arrangements  B Same chemical composition and different structural arrangements
	B Same chemical composition and different structural arrangements C Different chemical composition and Different structural arrangements D Different chemical composition and Same structural arrangements
1.1d	Hemocynin contain which metal?
	A Iron
	B . Copper C Silver
	D Gold
1.1e	Which is the not type of square planner complex?
	A MA <sub>4</sub>
	B MA <sub>3</sub> B C MAB <sub>3</sub>
	D M <sub>2</sub> A <sub>2</sub>
1,1f	PCl <sub>5</sub> + 2H <sub>2</sub> O
	A dichloro sulphuric acid
	B dichloro phospho acid C dichloro phosphoric acid
	D Trichloro phosphoric acid
1.1g	The ferritin has not been combined with iron is commonly termed as.
	A apoferritin B aboferritin
	C acoferritin
	D adoferritin
1.1h	Chemical formula of Boroxine
	A (BH) <sub>3</sub> O <sub>3</sub>
	B (BN) <sub>3</sub> H <sub>6</sub> C (BH) <sub>3</sub> O <sub>6</sub>
	C (BH) <sub>3</sub> O <sub>6</sub> D (BN) <sub>3</sub> O <sub>3</sub>
1.1i	Organometallic magnets were first discovered in the
	A 1982's
	B 1988's
	C 1980's D 1981's
1.1j	Body absorbed excessive amount of iron known the term as
	A Hemocromasosis

B

C

Hemobromatosis

Hemocromacosis D Hemocromatosis

1.2	Five Questions (MCQ/Short Question/Fill in the Blanks)	05
1.2		
1.21	b Definition: Condensation Polymer	
1.20	What is siderophores?	
1.20		
1.2	echelator used in Iron overload.	
Q.2	Short Notes (Attempt any two)	06
A	Chelation agents	
В	Polyphosphazenes	
C	Molar ratio method for Stability Constant	
Q.3	Explain in detail (Attempt any two)	14
Q.S A	Explain Write the notes on Cadmium toxicity (Occurrence, Health consideration	
11	and Toxic effect).	
В	Explain Factor that Influence Complex Formation.	
C	What is Ligand and Explain Types of Ligands.	
	A MA	
	RAM D	
	Five Guastians (MSQ /Sourc Question / pr	
	Divid the name of South was a common bulk	
	Anna Margama of It Clear of	
	The the water Education mare!	
	A State of the property of the state of the	
	- mirristos G	
	Compound	
	10c(M8) -Q	
	1.11 Deconometallic manages very legal equivalentation	
	Very Short Quantities 8	
	Very Short Questions Collectives	